Research Article　　　　　　　　　　　　　　　　　　**Open Access**

Regioselective C-H Borylation of C (sp²)-H Bond

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Abstract

C-H activation reactions have become a powerful method to direct functionalization of alkyl, alkenyl, and aryl C-H bonds over the past few decades. Among of them, Iridium catalyzed transformation of aryl C-H bonds to C-B bonds is one of the most useful method. However, a central challenge in these reactions is controlling their site selectivity. Over the past decade, some methods have been developed to accomplish regio-selective C-H borylation by catalysts or substrates modification. In this paper, some methods developed in recent years to realize ortho-, meta-, and para-selective C-H borylation will be summarized and their strategy and mechanism of these methods will be discussed.

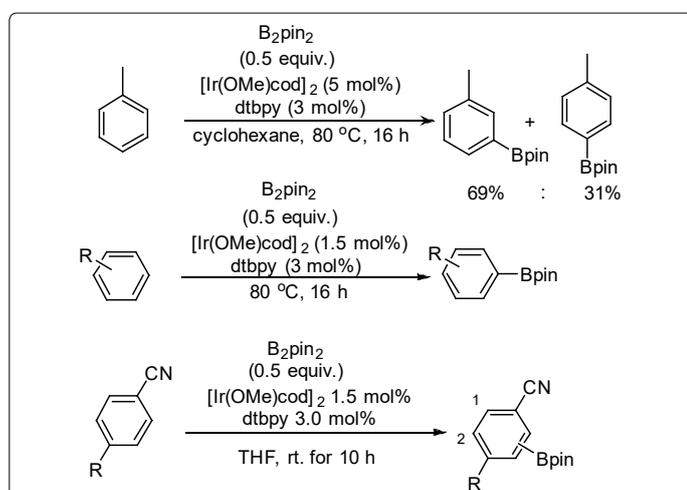**Keywords:** Iridium; Regioselective; C-H borylation; Non-covalent bond interaction.

Introduction

Carbon-Carbon bond are the molecular "bricks and mortar" from which diverse architectures in living organisms and manmade materials are constructed. In the field of organic chemistry, there are numerous methods for carbon-carbon bond construction, which have been developed, ranging from traditional nucleophilic reaction to metal catalyzed reaction for formation of C-C bond. Among of them, Suzuki-Miyaura reaction has become one of the most powerful method for the construction of C-C bond since its discovery in 1979 [1]. Arylboron is an important reagent for participating in this reaction. In addition, organoboron reagents as versatile intermediates have been extensively used in synthetic chemistry because it can be converted to more complex molecules by further transformations, such as Chan-Lam-Evans coupling, [2-4] and oxidation [5-7]. Therefore, developing a highly effective method for synthesis of organoboron reagents is very desirable.

Traditional methods to obtain organoboron reagents are these three methods (**Scheme 1**), which contains synthesis from organolithium or organomagnesium, hydroboration of alkenes or alkynes and haloboration of terminal alkynes [8]. However, there are some limitations of these methods. For example, the synthesis from organometallics usually requires stoichiometric amounts of strong base such as *n*-BuLi and a harsh lower temperature to use in the reaction. As for the reactions of hydroboration and haloboration, boron atom always adds to the terminal position of alkenes or alkynes. It is difficult to obtain internal boron reagents form these two methods. Besides, these three methods also are not in accordance with the rule of atomic economy. The development of transition metal catalyzed C-H borylation provides a good way for preparation of organoboron reagents, especially for aromatic boron reagents.
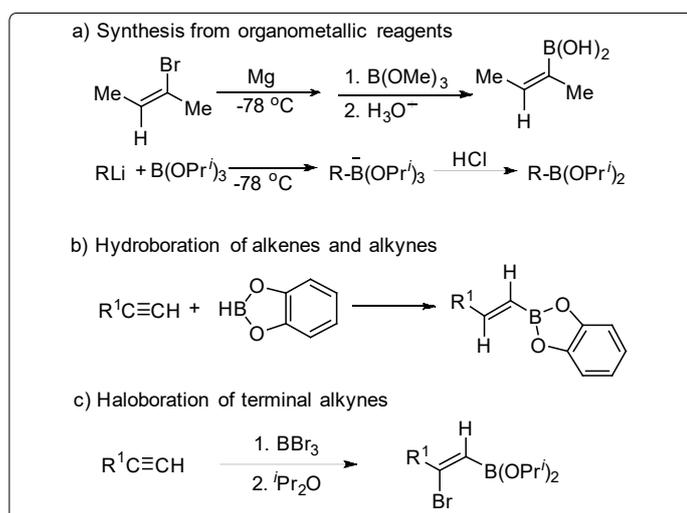
**Scheme 1.** Traditional methods for synthesis of organoboron reagents.

Scheme 2. Steric effect controls regioselectivity.

The most active catalyst for the borylation of aromatic compounds is the iridium/bipyridine catalytic system, which was found by Prof. Hartwig, Ishiyama, Miyaura et al. in 2002 [9-11]. This iridium catalytic system could make the C-H borylation of aromatic compounds proceed easily under mild condition. Moreover, they also found the C-H borylation reaction often occurs with regioselectivity controlled predominantly by steric effect [11] (**Scheme 2**). For instance, *mono*-substituted arenes as substrates, borylation always gives a mixture of *meta*- and *para*-borylated products in 2:1 ratio [9]. The *ortho*-borylated isomer was usually not formed because the steric hindrance of methyl substituent. In the case of *di*-substituted arenes, borylation reaction always proceeds at the position with less steric hindrance [12,13]. Reactions of 1, 4-disubstituted aromatic compounds could account for this steric effect more apparently. Borylation of asymmetrically 1, 4-disubstituted substrates, borylation proceeds at the *ortho*-position of substituent with less steric hindrance between the two possible reactive positions. Although steric effect could control regioselectivity of this reaction, achieving accurately control site-selectivity still exist many challenges in this kind of reaction. In the following, some methods developed in recent years will be summarized according to *ortho*-, *meta*- and *para*-selectivity.
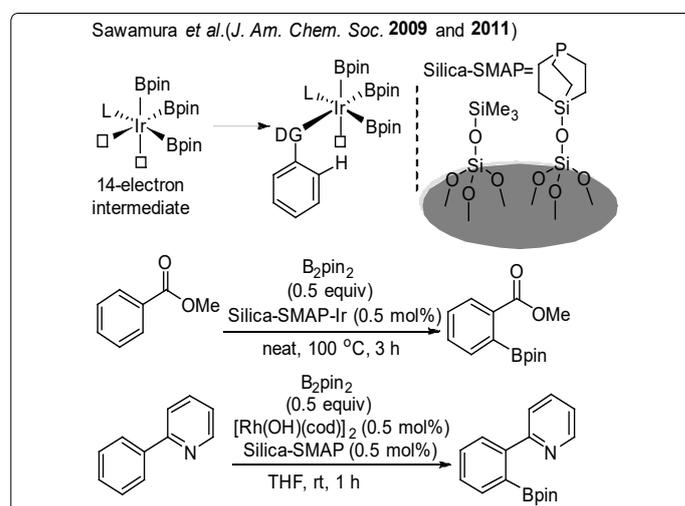
Ortho-Selective C-H Borylation

Directing group assist *ortho*-selective C-H borylation was firstly developed by Boebel et al. in 2008 [14]. They disclosed the example of *ortho*-selective C-H borylation for aromatic substrates directed by alkylhydrosilyl group (**Scheme 3**). This reaction proceeded according to a relay directed process. The alkylhydrosilyl group could reversibly attach to iridium center by σ-bond metathesis process to form a 16-electron intermediate, and bring iridium-boryl catalytic species close to *ortho*-position, which will facilitate the cleavage of ortho C-H bond. Benzyl dimethylsilane as substrate, the reaction could proceed with high *ortho*-selectivity in good yield.
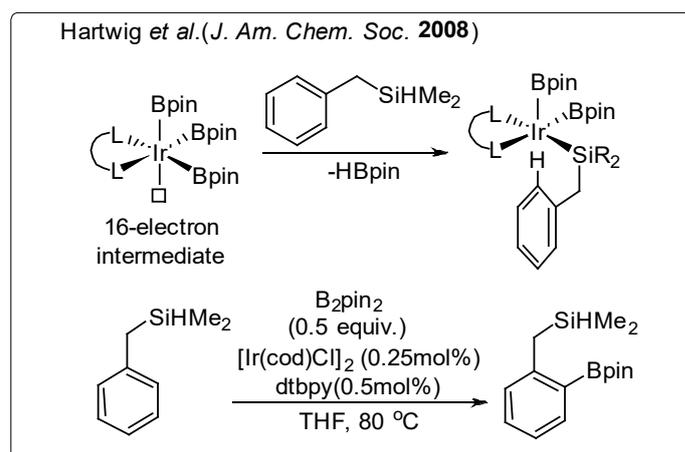

Scheme 3. Alkylhydrosilyl group directed *ortho*-selectivity.

In 2009, Kawamorita et al. [15,16] reported an *ortho*-selective C-H borylation of aromatic compounds catalyzed by silica supported *mono*-dentate phosphine ligand. In this reaction, *mono*-dentate ligand and iridium formed a 14-electrons intermediate, which contains two vacant sites, one for directing group of substrate and the other for the cleavage of ortho C-H bond (**Scheme 4**). Therefore, the *ortho*-selective C-H borylation proceeded through the intermediate. The reaction has very wide substrates scope including benzoate, ether, sulfonate and so on. In addition, he also developed an *ortho*-selective C-H borylation of 2-phenylpyridine in 2011, which was catalyzed by Rh with silica supported *mono*-dentate phosphine ligand.
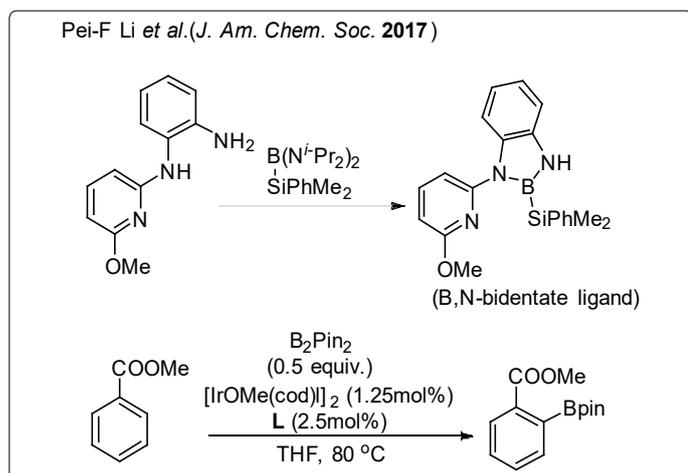

Scheme 4. *mono*-Dentate ligand controlled *ortho*-selectivity.

Later, Ligand-enabled *ortho*-selective C-H borylation was found by several groups, in which *ortho*-selectivity is achieved by modifying ligand structure (**Scheme 5**). In traditional methods, bipyridine type ligand is the most commonly used in iridium catalyzed sp^2 C-H borylation reaction. In 2014, Ghaffari et al. [17] reported silyl-phosphorous ligand catalyzed *ortho*-selective C-H borylation of alkyl benzoate. The reaction not only gave good yields of borylated product but also controlled *ortho*-selectivity very well. In 2016, Bisht and Chattopadhyay [18] also developed *ortho*-selective C-H borylation of benzaldehydes using 8-aminoquinoline as ligand, in which *tert*-butylamine was as the traceless protecting/directing group [17,18]. These two *ortho*-borylations could be suitable for a broad range of substrates and all of them give good to excellent yield and regioselectivity. The mechanism of these two reactions is similar to directing group assisting process. Designed ligands occupy two vacant orbital's, directing group brings the other one close to *ortho*-position.
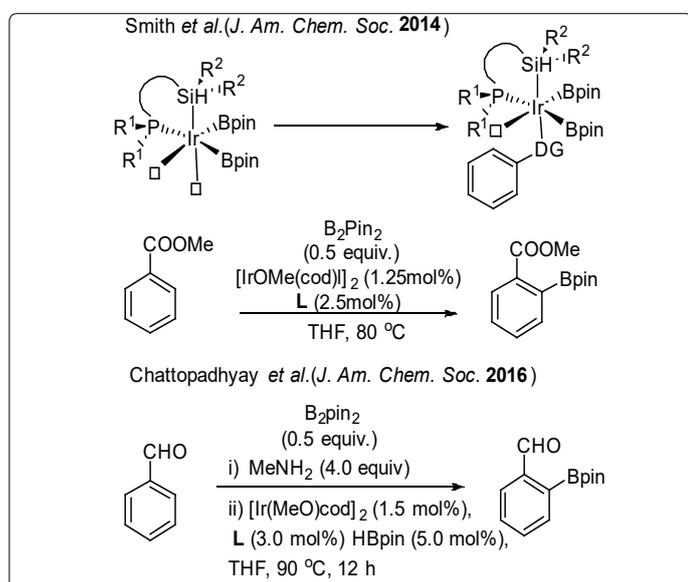
Scheme 5. Bidentate ligand enabled *ortho*-selectivity.

In addition, Wang et al. [19] reported a new *ortho*-selective C-H borylation catalyzed by a designed N, B-bidentate boryl ligand in 2017. Introducing convenient silylborane precursors onto N, B-bidentate boryl ligands, the iridium (III) complex was formed via Si-B oxidative addition.

Scheme 6. N, B-Bidentate ligand controlled *ortho*-selectivity.

However, no matter in directing group assist *ortho*-borylation or in ligand-enabled *ortho*-borylation, all of them usually require already presence or installation of a directing group in substrates. In contrast, traceless directing group would be more attractive alternatives; non-covalent bond interaction controlled regioselective C-H borylation reaction is a new strategy. At present, non-covalent organocatalysis has been successfully applied into some reactions to achieve regioselective C-H borylation of aromatic substrates by employing hydrogen bonding, ion pairing, Lewis acid-base interaction and electrostatic interaction. The first *ortho*-selective C-H borylation of aromatic compounds controlled by non-covalent bond interaction was found by Smith et al. in 2012, in which hydrogen bonding interaction between the H atom of Boc protected aniline substrate and the O atom of Bpin ligand favored *ortho*-selective C-H borylation [20].
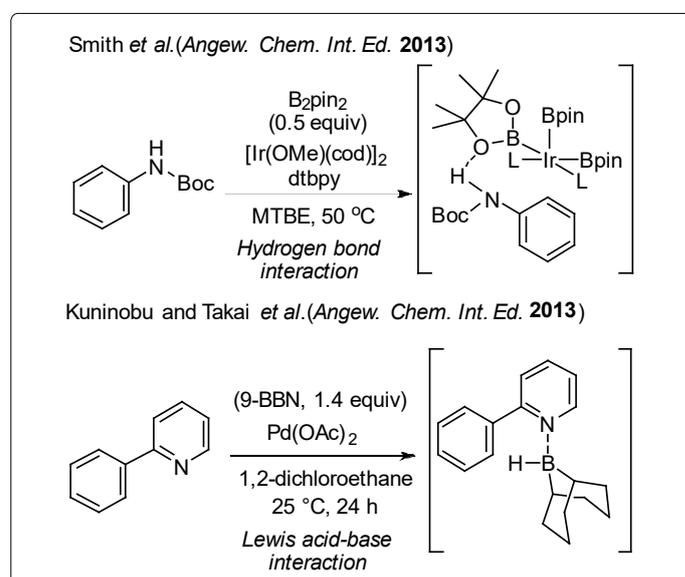
Scheme 7. Non-Covalent bond interaction controlled *ortho*-selectivity (I).

In 2013, Kuninobu [21] reported an *ortho*-selective C-H borylation of 2-phenylpyridine. In this reaction, 9-borabicyclo [3.3.1] nonane (9-BBN) was selected as boryl reagent; the boryl group was introduced at the *ortho*-position of 2-phenyl pyridine by Lewis acid-base interaction between the Lewis basic N atom and the Lewis acidic B atom [20]. Based on this result, Li et al. [22] developed an *ortho*-selective borylation of aryl sulfides using a Lewis acid-base interaction between designed ligand in the catalyst and a substituent (a sulfur atom) of the substrates in 2017 (**Scheme 8**). They think the steric repulsion between the iridium-boryl catalytic species and substituent(s) of the substrates would be an obstacle to realize *ortho*-selectivity; it is difficult to promote the site-selectivity using other non-covalent bond interaction such as hydrogen bonding interaction. Thus, a stronger interaction was used in this reaction. Almost the same time; Smith et al. reported a strategy for *ortho*-selective borylation of phenol derivatives. From selectivity of observation with ArylOBpin (pin = pinacolate) [23], they hypothesized that an electrostatic interaction between the partial negatively charged OBpin group and the partial positively charged bipyridine ligand of the catalyst favors *ortho*-selectivity.

Scheme 8. Non-Covalent bond interaction controlled *ortho*-selectivity (ll).
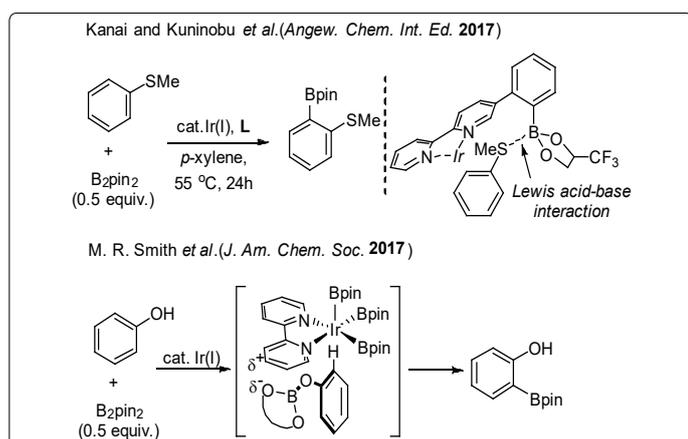
meta-Selective C-H Borylation

The small difference in intrinsic reactivity of C-H bonds in organo-molecules makes it difficult to achieve regiocontrol. Directing group assist *ortho*-selectivity by forming five or six cyclometallation has been developed very well in the past years. However, the directed activation of C-H bond that are distal to directing group still exist challenges such as *meta*- and *para*-selectivity because the target C-H bond is geometrically inaccessible to directed metalation owing to the ring strain. In this context, non-covalent bond interaction between catalyst species and substrate provide a good way to solve this problem. Recent years, *meta*- and *para*-selective C-H borylation of aromatic compounds have been achieved using non-covalent bond interaction. A significant breakthrough was made by Kuninobu group in 2015 [24]. They developed an iridium-catalyzed *meta*-selective C–H borylation of aromatic compounds using a newly designed catalytic system (**Scheme 9**). The hydrogen bonding interaction between the urea moiety of designed ligand and a hydrogen-bond acceptor in the substrate (carbonyl of amide) places the iridium catalyst to the *meta*-position of aromatic amides, eaters, phosphonates, and phosphonic diamide and phosphine oxides. When compared with directing group-controlled reaction, the hydrogen-bonding ligand only required in a catalytic amount which interacts reversibly with the substrate.


Scheme 9. Hydrogen-bonding interaction controlled *meta*-selectivity.
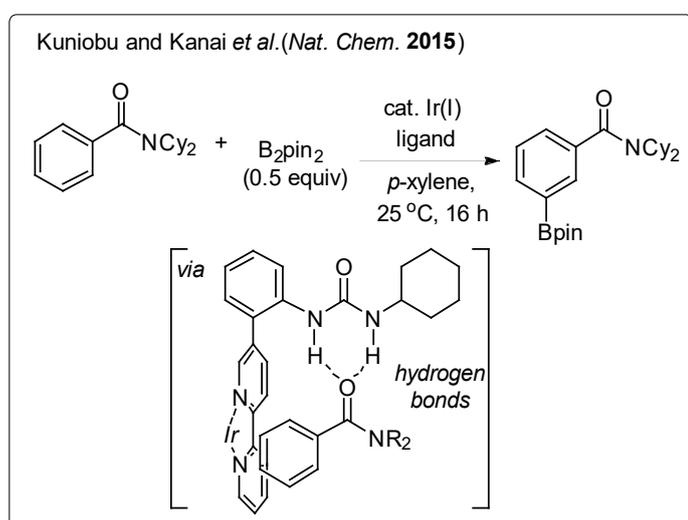
In 2016, Davis et al. reported an ion pair-directed approach to controlling regioselectivity in the iridium-catalyzed borylation of two classes of aromatic quaternary ammonium salts [25]. A single electrostatic interaction could be successfully employed to position a reactive metal catalytic species to *meta*-target C-H bond. This is the first example to demonstrate the viability of ion-paring as a powerful tool for region control.

Scheme 10. Ion-paring and Lewis acid-base interaction controlled *meta*-selectivity
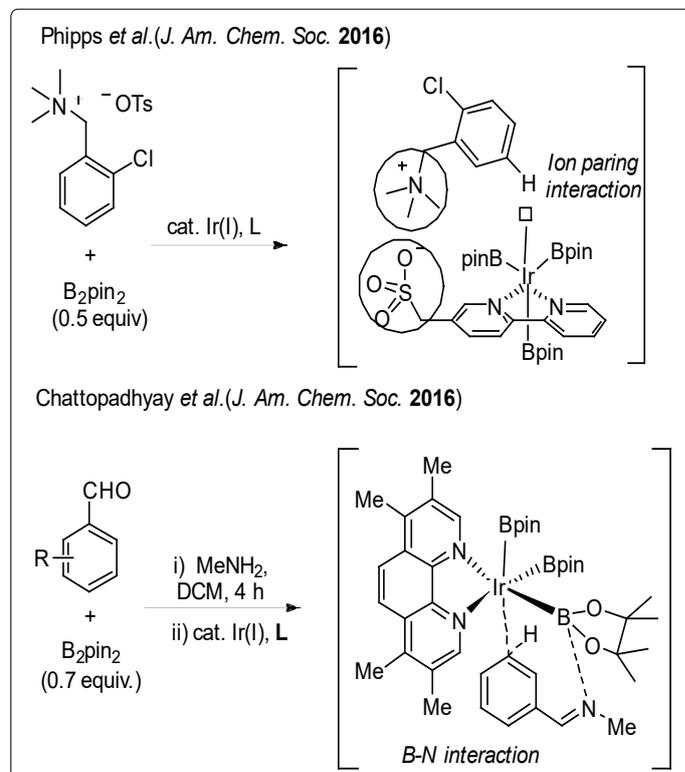
Besides, Bisht reported *meta* borylation proceeds via an electrostatic interaction and a secondary interaction between a designed ligand and substrate [18]. The origin of *meta*-selectivity was controlled by two factors: i) an electrostatic interaction of the tris(boryl)iridium complex attached with the electron-rich ligand substrate; ii) a secondary Lewis acid-base interaction between the imine N atom and the boryl B atom of catalyst.
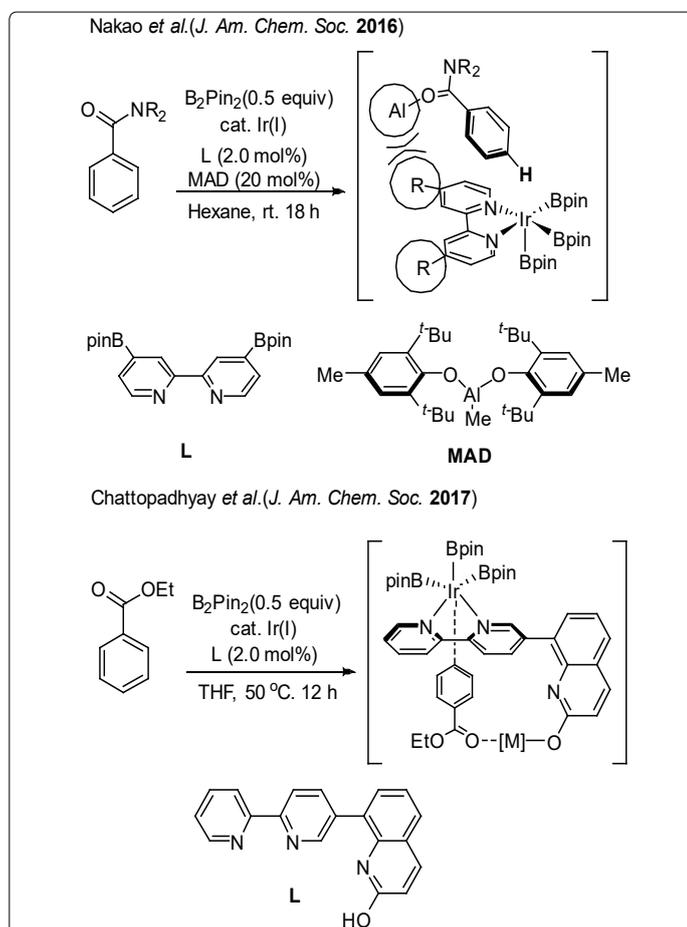
para-Selective C-H Borylation

Electronic aromatic substitution is a traditional method to achieve *para*-selective C-H functionalization. However, the substrate scope of this method is very limit, only when strongly electron-donating groups (EDG) such as a dimethylamino group are attached to the benzene rings. In addition, it always gives a mixture of *ortho*- and *para*-product at the same time. Therefore, many groups were devoted into developing a strategy that only favors *para*-selective functionalization. In 2015, Saito accomplished a highly *para*-selective aromatic C-H borylation of mono-substituted benzene derivatives by using a new iridium catalyst bearing a bulky diphosphine ligand. The site-selectivity raised from the steric repulsion between the bulkiness of substituent on benzene ring and the diphosphine-iridium catalyst [26].
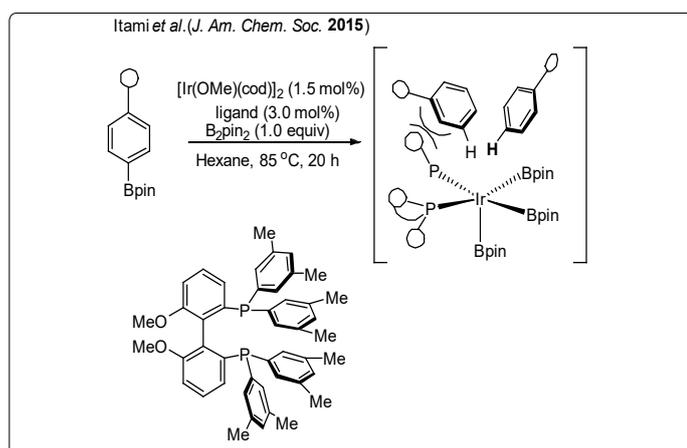
Scheme 11. Steric effect controlled *para*-selectivity.

By using the strategy of steric repulsion between catalyst and substrate, Yang and coworkers reported a method of *para*-selective C-H borylation of benzamides in 2017 by using cooperative iridium/aluminum catalysis [27]. They thought that the regioselectivity is controlled by the steric repulsion between the coordination of substrate to the bulky aluminum and the boryl-iridium catalytic species because the coordination shields the *ortho*- and *meta*- reactive position of substrate (**Scheme 11**). In addition, Hoque disclosed to a highly efficient method for *para*-selective borylation [28]. By modifying the core structure of bipyridine, the designed L-shape ligand will recognize the functionality of the oxygen atom of the ester carbonyl group via non-covalent interaction, which provides an unprecedented controlling factor for *para*-selective C-H activation.

Scheme 12. Ligand enabled *ortho*-selective C-H borylation.

Conclusion

Organo-boron compounds as versatile intermediate has been widely employed in organic synthesis. The development of regioselective C-H borylation provides an efficient way to prepare them using different strategies such as directing group assist *ortho*-selectivity, ligand enable *meta*- and *para*-selectivity. However, there are many challenges that remain to be addressed to improve the practicality and versatility of C-H borylation reaction in the future. For instance, it will be of great importance to develop more effective ligand that enable regioselective C (sp^3)-H borylation, especially for remote position. C (sp^3)-H bond is building block for constructing natural products, regioselective functionalization could highly improve the efficiency for preparing biological molecules.

Acknowledgment

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